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Name.....

Reg. No.....

FIRST SEMESTER M.Sc. DEGREE EXAMINATION, DECEMBER 2016

(CUCSS)

Chemistry

CH 1C 01—QUANTUM CHEMISTRY AND GROUP THEORY

(2015 Admissions)

Time : Three Hours

Maximum : 36 Weightage

Part A

Answer all questions.

Each question carries a weightage of 1.

1. An electron is **confined** to one-dimensional box of length 10 nm. Find the uncertainty in momentum.
2. Which of the following are well behaved functions ? Justify your answer :

(a) e^x

(b) e^{ix}

(c) e^{-ax^2}

(d) \sin^{-1}

3. How does the concept of degeneracy arise in quantum mechanical problems ?
4. Write Hamiltonian for (a) H_e ; (b) H_2 .
5. Write Rodrigue's formula. Explain.
6. Where do you find maximum electron density for is wave function ? Justify your answer.
7. Explain spin functions and spin operators.
8. Write the examples for spherical harmonics.
9. Assign Schoenflies symbol of point group for the following :
 - (a) Cyclohexane (boat form).
 - (b) Allene.
10. Distinguish between inverse and conjugate operations with examples.
11. Explain the term 'Gamma cart'.
12. List symmetry operations possible as D_{3h} . Classify them into different classes of operations.

(12 x 1 = 12 weightage)

Turn over

Part B

Answer any eight questions.

Each question carries a weightage of 2.

13. Show that if the operators commute they will have the same set of eigen functions and eigen values.
14. Show that \hat{L}^2 commutes with L.
15. Find the average value of position of a particle confined to one-dimensional box of length a
 $\psi_n = \sqrt{2/a} \sin(\pi/a)x$.
16. Apply Schrödinger wave equation for a planar rotor. Find eigen functions and eigen values.
17. The is wave function is given as $\frac{1}{\sqrt{a_0}} \left(\frac{r}{a_0}\right)^{3/2} e^{-r/a_0}$. Show that the maximum probability of find the electron is at $r = a_0$.
18. Draw polar diagrams for 3d atomic orbitals. Discuss.
19. State and explain expectation value postulate.
20. What is block **diagonalization** ? Explain its significance.
21. Use Great Orthogonality theorem to show that 'sum of the squares of the characters of any representation is the order of the group'.
22. Derive C_3 character table.
23. Taking the positional co-ordinates of all the atoms in ethylene (D_{2h}) derive a reducible representation (write only characters of the corresponding matrices). The symmetry operations are ;
 E, C₂, C₂, C₂, σ_h , σ_v , i .
24. Generate group multiplication table for C_{3v}

(8 x 2 = 16 weightage)

Part C

Answer any two questions.

Each question carries a weightage of 4.

25. Apply **Schrödinger** wave equation for a simple harmonic oscillator. Find **eigen** functions and **eigen** values.
26. Write Great Orthogonality theorem. What are the consequences of the theorem? Use the theorem to derive reduction formula.
27. Discuss the systematic procedure to assign **Schoenflies** symbol of point group.
28. Discuss briefly :
 - (a) Ladder operator.
 - (b) **Legendre** polynomials.
 - (c) Postulate of spin by **Uhlenbeck** and **Goudsmith**.

(2 x 4 = 8 weightage)