

D 93042

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Name.....

Reg. No.....

FIRST SEMESTER M.Sc. DEGREE EXAMINATION, DECEMBER 2015

(CUCSS)

Chemistry

CH 1C 04—THERMODYNAMICS, KINETICS AND CATALYSIS

(2015 Admissions)

Maximum : 36 Weightage

Time : Three Hours

Part A

Answer all questions.

Each question carries 1 weightage.

1. What is residual entropy ? How it differs from excess entropy ?
2. What is the need of third law of thermodynamics ?
3. Define thermal diffusion.
4. Write and explain Glansdorf-Pregogine equation.
5. What is explosion limit ? Explain with an example.
6. What is secondary salt effect ? Explain with an example.
7. Write the Eyring equation and write its significance.
8. What are crossed molecular beams ? Write their uses.
9. Write BET equation and explain the terms involved.
10. Distinguish between SEM and TEM.
11. What is Van't Hoff and Arrhenius intermediates ?
12. Define autocatalysis with example.

(12 × 1 = 12 weightage)

Part B

Answer any eight questions.

Each question carries 2 weightage.

13. What are the apparent exceptions to third law of thermodynamics ? Write its applications.
14. Explain how the non-ideal solutions deviate from Raoult's law.
15. Explain entropy production.
16. Explain thermo-osmosis with example.
17. Write the principle of relaxation technique.

Turn over

18. What is the effect of dielectric constant on reaction rate ?
19. Write the assumptions of collision theory.
20. What are attractive and repulsive surfaces ?
21. Distinguish between Langmuir and BET adsorption isotherms.
22. How will you determine surface area from BET equation.
23. Explain specific acid catalysis with example.
24. What are oscillating reactions ? Give an example.

(8 × 2 = 16 weightage)

Part C

*Answer any two questions.
Each question carries 4 weightage.*

25. Derive Duhem-Margules equation and write its applications.
26. Discuss the kinetics of chain reactions.
27. Write RK theory of unimolecular reactions.
28. Derive Michaeli's-Menton equation and explain its significance.

(2 × 4 = 8 weightage)