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FIFTH SEMESTER B.A./B.Sc. DEGREE EXAMINATION **NOVEMBER 2020**

(CUCBCSS—UG)

Chemistry

CHE 5B 07—ORGANIC CHEMISTRY—II

Time · Three Hours		Maxi	mum: 80 Mark
Time : Inree nours		21224	

Section A

	Answer all questions. Each question carries 1 mark.
1.	What is the product obtained when nitrobenzene is reduced electrolytically in strong acidic medium?
2.	Scientific study of fermentation was first made by ———.
3.	Arrange in the order of increasing base strength – $C_6H_5NH_2$, $C_6H_5CH_2NH_2$, $C_6H_5NHCH_3$.
4.	Active methylene compounds react with aldehydes in the presence of pyridine to give α , β -unsaturated acids. This reaction is known as ———.
5.	Zeisel method is used to estimate — group in an organic compound.
6.	Phenol can be distinguished from ethanol by reaction with ———.

- 7. Chlorobenzene on heating with NaOH at 300° C under pressure gives -
- 8. Which of the following will give iodoform test?
 - (a) CH₃OH;

- (b) CH₃CH (OH) CH₃;
- (c) $CH_3CH_2CH_2OH$; and (d) $C_6H_5CH_2OH$.
- 9. Identify the products in the following reaction.

2 PhCHO ____:OH

10. What is the product obtained when anisole is treated with HI?

 $(10 \times 1 = 10 \text{ marks})$

Turn over

Section B

Answer at least five questions. Each question carries 4 marks. All questions can be attended. Overall Ceiling 20.

- 11. 2, 4, 6-trinitrochlorobenzene can be easily hydrolysed with aqueous sodium carbonate. Why?
- 12. What is Nef reaction?
- 13. What is Walden inversion?
- 14. A liquid was mixed with ethanol and a drop of conc. H_2SO_4 was added. A compound with a fruity smell having molecular formula $C_4H_8O_2$ was formed. Identify the liquid and write down the reaction.
- 15. Compare the acidity of the following with suitable explanations:

- 16. Carbonyl compounds react with Grignard reagent to give almost all types of alcohols. Justify this statement.
- 17. Identify the reactions and write down the products X and Y in the following reaction.

$$H_3C$$
— CH_2 — $COOH$ — CI_2 X alc. KOH Y

- 18. Write a note on Lucas reagent.
- 19. How will you prepare benzene sulphonic acid?
- 20. Write a note on Jones reagent.
- 21. Write down the reagents used in Clemmenson and Wolff-Kishner reduction.
- 22. Explain role of lactic acid in exercise.

 $(5 \times 4 = 20 \text{ marks})$

Section C

Answer at least **four** questions. Each question carries 7.5 marks. All questions can be attended. Overall Ceiling 30.

- 23. What will be the product when propionaldehyde is treated with NaOH? Discuss the mechanism involved in the reaction.
- 24. Write down the product and depict the mechanism involved in the following conversion.

- 25. Write a note on colour change of phenolphthalein with pH.
- 26. Give any two applications of organozinc compounds and ethylacetoacetate.
- 27. How does Saytzeff's rule in elimination differ from Hofmann's elimination? Illustrate with suitable examples.
- 28. What is Hinsberg reagent? Write down its importance in organic chemistry.
- 29. Write a note on: (a) Explosives; (b) Decarboxylation; and (c) Preparation and use of vanillin.
- 30. Identify X, Y and Z. Among the conversions which is diazotization process? Explain.

$$\begin{array}{c|c}
 & \text{NH}_2 \\
\hline
& Br_2 \text{ (aq)} \\
\hline
& X &
\end{array}$$

$$\begin{array}{c|c}
 & \text{NaNO}_2 + \text{HCI} \\
& Y &
\end{array}$$

$$\begin{array}{c|c}
 & \text{H}_3 \text{PO}_2 + \text{H}_2 \text{O} \\
& \end{array}$$

$$\begin{array}{c|c}
 & \text{Z}
\end{array}$$

 $(4 \times 7\frac{1}{2} = 30 \text{ marks})$

Turn over

Section D

Answer any **two** questions. Each question carries 10 marks.

- 31. Give a detailed account of elimination- addition and addition elimination mechanism in aromatic nucleophilic substitution reactions. Illustrate with suitable examples.
- 32. Write down the mechanism of the following reactions:
 - (a) Riemer-Tiemann reaction; (b) Perkin's reaction; and (c) Claisen rearrangement.
- 33. Explain the following reactions: (a) Swarts reaction; (b) Schmidt reaction; (c) Etard reaction; (d) Carbylamine reaction; and (e) Claisen condensation.
- 34. Explain the following: (a) Basicity of Guanidine; (b) Preparation and uses of sulpha drugs; and (c) Preparation and uses of citric acid.

 $(2 \times 10 = 20 \text{ marks})$